

Chapter 18: Electrical Properties

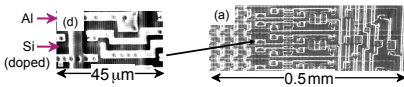
ISSUES TO ADDRESS...

- How are electrical conductance and resistance characterized?
- What are the physical phenomena that distinguish _____?
- For metals, how is _____ affected by _____ and deformation?
- For semiconductors, how is conductivity affected by impurities (doping) and temperature?

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View of an Integrated Circuit

- Scanning _____ micrographs of an IC:



- A dot map showing location of Si (a semiconductor):
-- Si shows up as _____ regions.




- A dot map showing location of Al (a conductor):
-- Al shows up as _____ regions.



Fig. (d) from Fig. 12.27(a), Callister & Rethwisch 3e. (Fig. 12.27 is courtesy Nick Gorzales, National Semiconductor Corp., West Jordan, UT.)

Figs. (a), (b), (c) from Fig. 18.27, Callister & Rethwisch 8e.

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Electrical Conduction

- _____ Law: $V = IR$
voltage drop (volts = J/C) \rightarrow resistance (Ohms)
 $C =$ _____ current (amps = C/s)

- _____, ρ :
-- a material property that is independent of sample size and geometry

$$\rho = \frac{RA}{L}$$

surface area of current flow (RA)
current flow path length (L)

- _____, σ

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Electrical Properties

- Which will have the greater _____?

$R_1 = \frac{2\rho l}{\pi\left(\frac{D}{2}\right)^2} = \frac{8\rho l}{\pi D^2}$

$R_2 = \frac{\rho l}{\pi\left(\frac{2D}{2}\right)^2} = \frac{\rho l}{\pi D^2} = \frac{R_1}{8}$

- Analogous to flow of water in a pipe
- _____ depends on sample geometry and size.

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Definitions

Further definitions

<= another way to state Ohm's law

$J = \text{current density} = \frac{\text{current}}{\text{surface area}} = \frac{I}{A}$ like a flux

$\epsilon =$ _____

$J = \sigma (V/l)$

Electron flux conductivity _____

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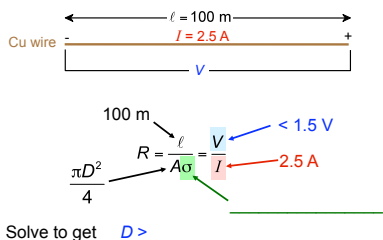
Conductivity: Comparison

- Room temperature values $(\text{Ohm}\cdot\text{m})^{-1} = (\Omega \cdot \text{m})^{-1}$

METALS	conductors													
Silver	6.8×10^7	Soda-lime glass 10^{-10} - 10^{-11}												
Copper	6.0×10^7	Concrete 10^{-9}												
Iron	1.0×10^7	Aluminum oxide $<10^{-13}$												
<table border="0" style="width: 100%;"> <tr> <td style="width: 30%;"></td> <td style="width: 30%;">POLYMERS</td> <td style="width: 30%;"></td> </tr> <tr> <td>Silicon</td> <td>4×10^{-4}</td> <td>Polystyrene $<10^{-14}$</td> </tr> <tr> <td>Germanium</td> <td>2×10^0</td> <td>Polyethylene 10^{-15}-10^{-17}</td> </tr> <tr> <td>GaAs</td> <td>10^{-6}</td> <td></td> </tr> </table>				POLYMERS		Silicon	4×10^{-4}	Polystyrene $<10^{-14}$	Germanium	2×10^0	Polyethylene 10^{-15} - 10^{-17}	GaAs	10^{-6}	
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semiconductors Selected values from Tables 18.1, 18.3, and 18.4, Callister & Rethwisch 8e.														

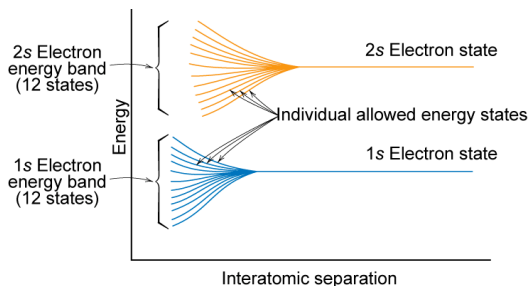
Example: Conductivity Problem

What is the minimum diameter (D) of the wire so that $V < 1.5$ V?



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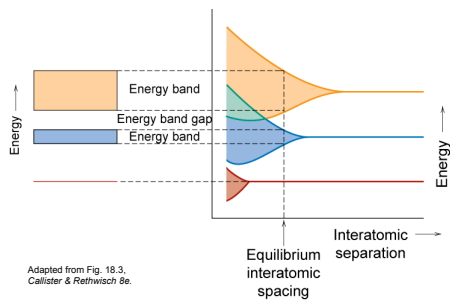
Electron Energy Band Structures



Adapted from Fig. 18.2, Callister & Rethwisch 8e.

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Band Structure Representation



Adapted from Fig. 18.3, Callister & Rethwisch 8e.

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Conduction & Electron Transport

- Metals (_____):
- for metals _____ are adjacent to filled states.
- thermal energy excites _____ into empty higher energy states.
- two types of band structures for metals:
 - empty band that overlaps filled band

Partially filled band

Overlapping bands

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Energy Band Structures: Insulators & Semiconductors

- _____:
- wide band gap (____ eV)
- few electrons excited across band gap
- _____:
- narrow band gap (____ eV)
- more electrons excited across band gap

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Metals: Influence of Temperature and Impurities on Resistivity

- _____ increases resistivity
- grain boundaries
- impurity atoms

These act to scatter electrons so that they take a less direct path.

- Resistivity _____ with:
 - temperature
 - wt% impurity
 - %CW

$$\rho = \rho_{\text{thermal}} + \rho_{\text{impurity}} + \rho_{\text{deformation}}$$

Adapted from Fig. 18.8, Callister & Rethwisch 8e (Fig. 18.8 adapted from J.O. Linde, Ann. Physik 5, p. 219 (1932); and C.A. Wert and R.M. Thomson, Physics of Solids, 2nd ed., McGraw-Hill Book Company, New York, 1970.)

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Estimating Conductivity

- Question:
 - Estimate the electrical conductivity σ of a Cu-Ni alloy that has a yield strength of 125 MPa.

Adapted from Fig. 7.16(b), Callister & Rethwisch 8e.

Adapted from Fig. 18.9, Callister & Rethwisch 8e.

From step 1:

$C_{Ni} = \underline{\hspace{2cm}} Ni$

$\sigma = \frac{1}{\rho} = 3.3 \times 10^6 (\text{Ohm} - \text{m})^{-1}$

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Charge Carriers in Insulators and Semiconductors

Adapted from Fig. 18.6(b), Callister & Rethwisch 8e.

Two _____ carriers:

- _____ charge
- in conduction band
- _____ charge
- vacant electron state in the valence band

Move at different speeds - drift velocities

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Intrinsic Semiconductors

- Pure material _____: e.g., silicon & germanium
 - Group IVA materials
- Compound semiconductors
 - _____ compounds
 - Ex: GaAs & InSb
 - _____ compounds
 - Ex: CdS & ZnTe
 - The wider the electronegativity difference between the elements the wider the energy gap.

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Intrinsic Semiconduction in Terms of Migration

- Concept of _____:
 - valence electron
 - Si atom
 - electron pair creation
 - electron = hole
 - hole pair migration

- Electrical _____ given by:
 - # electrons/m³
 - # holes/m³
 - electron mobility
 - hole mobility

$$\sigma = n|e|\mu_e + p|e|\mu_h$$

Adapted from Fig. 18.11, Callister & Rethwisch 8e.

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Number of Charge Carriers

Conductivity

$$\sigma = n|e|\mu_e + p|e|\mu_h$$

- for _____ semiconductor $n = p = n_i$
- $\therefore \sigma = n_i|e|(\mu_e + \mu_h)$
- Ex: GaAs

$$n_i = \frac{\sigma}{|e|(\mu_e + \mu_h)} = \frac{10^{-6} (\Omega \cdot m)^{-1}}{(1.6 \times 10^{-19} C)(0.85 + 0.45 m^2/V \cdot s)}$$

For GaAs $n_i =$ _____

For Si $n_i = 1.3 \times 10^{16} m^{-3}$

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Intrinsic Semiconductors: Conductivity vs T

- Data for _____:
 - σ increases with T
 - opposite to metals

$$n_i \propto e^{-E_{gap}/kT}$$

material	band gap (eV)
Si	1.11
Ge	0.67
GaP	2.25
CdS	2.40

Selected values from Table 18.3, Callister & Rethwisch 8e.

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Intrinsic vs Extrinsic Conduction

- case for pure Si
 -- # electrons = # _____ ($n = p$)
- electrical behavior is determined by presence of impurities that introduce excess electrons or holes
- n -type _____ : ($n \gg p$) • p -type Extrinsic: ($p \gg n$)

Phosphorus atom

no applied electric field

Boron atom

no applied electric field

Adapted from Figs. 18.12(a) & 18.14(a), Callister & Rethwisch 8e. Chapter 18 - 19

Extrinsic Semiconductors: Conductivity vs. Temperature

- Data for _____ :
 -- reason: imperfection sites lower the activation energy to produce mobile electrons.
- Comparison: _____ vs extrinsic conduction...
 -- doping level: $10^{21}/m^3$ of a n -type donor impurity (such as P).
 -- for $T < 100$ K: "freeze-out", thermal energy insufficient to excite electrons.
 -- for 150 K $< T < 450$ K: "extrinsic"
 -- for $T >> 450$ K: "intrinsic"

Adapted from Fig. 18.17, Callister & Rethwisch 8e. (Fig. 18.17 from S.M. Sze, Semiconductor Devices, Physics, and Technology, Bell Telephone Laboratories, Inc., 1965.)

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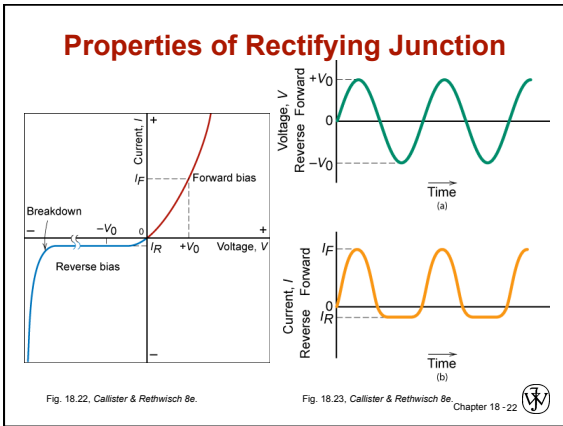
p-n Rectifying Junction

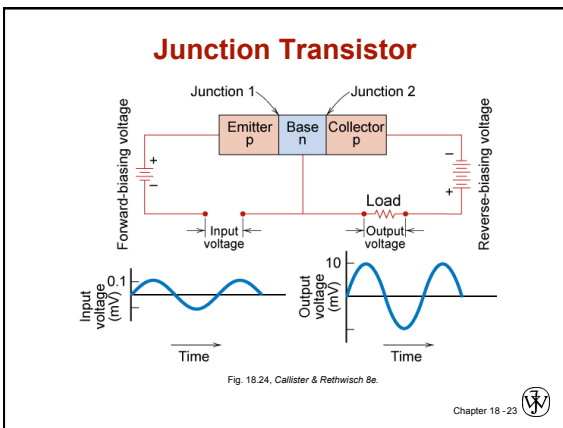
- Allows flow of _____ in one direction only (e.g., useful to convert _____ current to _____ current).
- Processing: diffuse P into one side of a B-doped crystal.

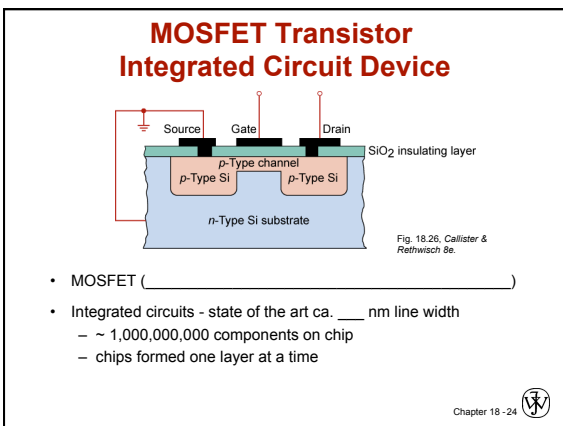
- No applied _____ :
 _____ bias: carriers flow through p -type and n -type regions; holes and electrons recombine at p - n junction; current flows.
- _____ bias: carriers flow away from p - n junction; junction region depleted of carriers; little current flow.

Adapted from Fig. 18.21 Callister & Rethwisch 8e.

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Ferroelectric Ceramics

- Experience _____ polarization

BaTiO₃ -- ferroelectric below its temperature (120°C)

● Ti⁴⁺ ● Ba²⁺ ● O²⁻

Fig. 18.35, Callister & Rethwisch 8e. Chapter 18 - 25

Piezoelectric Materials

- application of stress induces _____
- application of voltage induces dimensional change

stress-free with applied stress

Adapted from Fig. 18.36, Callister & Rethwisch 8e. (Fig. 18.36 from Van Vlack, Lawrence H., Elements of Materials Science and Engineering, 1988, p.482, Adapted by permission of Pearson Education, Inc., Upper Saddle River, New Jersey) Chapter 18 - 26

Summary

- Electrical *conductivity* and *resistivity* are:
 - material parameters
 - geometry independent
- Conductors, semiconductors, and insulators...
 - differ in range of conductivity values
 - differ in availability of electron excitation states
- For metals, *resistivity* is increased by
 - increasing temperature
 - addition of imperfections
 - plastic deformation
- For pure semiconductors, *conductivity* is increased by
 - increasing temperature
 - doping [e.g., adding B to Si (*p*-type) or P to Si (*n*-type)]
- Other electrical characteristics
 - ferroelectricity
 - piezoelectricity

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