**Supporting Information for** 

# **Redox-active, near-infrared dyes based on 'Nindigo' (indigo-***N***,***N***'-diarylimine) boron chelate complexes**

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### <sup>13</sup>C NMR spectrum of **2a**









ppm

#### <sup>11</sup>B NMR spectrum of **2a**

-126.8

ī.

-127.0

-127.2







#### <sup>19</sup>F NMR spectrum of **2b**







<sup>1</sup>H NMR spectrum of **2c.** Peak at ~1.5 ppm is due to residual water and peak at ~5.2ppm due to solvent.





-100 -105 -110 -115 -120 -125 -130 -135 -140 -145 -150 -155 -160 -165 -170 -175 -180 -185 -190 -195 ppm

<sup>11</sup>B NMR spectrum of **2c** 





<sup>1</sup>H NMR spectrum of **2d.** Peak at ~1.5 ppm is due to residual water and peak at ~5.2ppm due to solvent.







<sup>13</sup>C NMR spectrum of **2e** 



<sup>19</sup>F NMR spectrum of **2e** 



<sup>11</sup>B NMR spectrum of **2e** 







#### <sup>19</sup>F NMR spectrum of **2f**



<sup>11</sup>B NMR spectrum of **2f** 



#### <sup>1</sup>H NMR spectrum of **3a**



## <sup>19</sup>F NMR spectrum of **3a**

2a (CDCl3)





## <sup>1</sup>H NMR spectrum of **3b**











<sup>11</sup>B NMR spectrum of **3c** 





Decomposition of BisBoroDmpNindigo after 52 hours in CD2C12 at room temperature





 $^{\rm 19}{\rm F}~{\rm NMR}$  spectrum of  ${\rm 3d}$ 



-100 -105 -110 -115 -120 -125 -130 -135 -140 -145 -150 -155 -160 -165 -170 -175 -180 -185 -190 -195 ppm

<sup>11</sup>B NMR spectrum of **3d** 



32051.281 15.600 6.50 297.4 1.00000000 1

f1

IANNEL f1 11B 14.10 usec -3.00 db 160.4616790 MHz 16384 160.4616000 MHz 0 1.80 Hz 0 1.40



BisBoroDippNindigo

1.8

<sup>1</sup>H NMR spectrum of **3f** 





<sup>11</sup>B NMR spectrum of **3f** 



use dB MHz MHz

#### **Electrochemical Data**



Cyclic voltammagram of 2b (CH<sub>2</sub>Cl<sub>2</sub> solution, 0.1 mM Bu<sub>4</sub>NBF<sub>4</sub> electrolyte and scan rate 100 mVs<sup>-1</sup>)



 $Cyclic \ voltammagram \ of \ 2d \ (CH_2Cl_2 \ solution, \ 0.1 \ mM \ Bu_4NBF_4 \ electrolyte \ and \ scan \ rate \ 100 \ mVs^{-1}$ 



Cyclic voltammagram of 2e (CH<sub>2</sub>Cl<sub>2</sub> solution, 0.1 mM Bu<sub>4</sub>NBF<sub>4</sub> electrolyte and scan rate 100 mVs<sup>-1</sup>



Cyclic voltammagram of 2f (CH\_2Cl\_2 solution, 0.1 mM Bu\_4NBF\_4 electrolyte and scan rate 100 mVs^{-1}  $\,$ 

## X-ray structures



ORTEP drawing of 2b (20% probability level). Aromatic hydrogen atoms are omitted for clarity



ORTEP drawing of the two crystallographically-independent molecules of 2c (20% probability level). Aromatic hydrogen atoms are omitted for clarity



ORTEP drawing of the two crystallographically-independent molecules of 2d (20% probability level). Aromatic hydrogen atoms are omitted for clarity



ORTEP drawing of 2f (20% probability level). Aromatic hydrogen atoms are omitted for clarity